

Material Safety Data Sheet

HAZARD WARNINGS	RISK PHRASES	PROTECTIVE CLOTHING
	The health risks of this compound have not been fully determined. Exposure may cause irritation of the skin, eyes, and respiratory system.	

Section I. Chemical Product and Company Identification			
Chemical Name	1,6-Dimethylnaphthalene		
Catalog Number	D0749	Supplier	TCI America 9211 N. Harborgate St. Portland OR 1-800-423-8616 Chemtrec® (800) 424-9300 (U.S.) (703) 527-3887 (International)
Synonym	Not available.		
Chemical Formula	$C_{10}H_6(CH_3)_2$		
CAS Number	575-43-9	In case of Emergency Call	
			(103) 321-3001 (International)

Section II. Composition and Information on Ingredients					
Chemical Name		CAS Number	Percent (%)	TLV/PEL	Toxicology Data
1,6-Dimethylnaphthalene		575-43-9	Min. 95.0 (GC)	Not available.	Not available.

Section III.	Hazards Identification	
Acute Health Effects	No specific information is available in our data base regarding the toxic effects of this material for humans. Howevexposure to any chemical should be kept to a minimum. Skin and eye contact may result in irritation. May be harm inhaled or ingested. Always follow safe industrial hygiene practices and wear proper protective equipment when hand this compound. Follow safe industrial hygiene practices and always wear proper protective equipment when handling this compound.	
Chronic Health Effects	CARCINOGENIC EFFECTS: Not available. MUTAGENIC EFFECTS: Not available. TERATOGENIC EFFECTS: Not available. DEVELOPMENTAL TOXICITY: Not available. Repeated or prolonged exposure to this compound is not known to aggravate existing medical conditions.	

Section IV.	First Aid Measures
Eye Contact	Check for and remove any contact lenses. In case of contact, immediately flush eyes with plenty of water for at least 15 minutes. Get medical attention.
Skin Contact	In case of contact, immediately flush skin with plenty of water. Remove contaminated clothing and shoes. Wash clothing before reuse. Thoroughly clean shoes before reuse. Get medical attention.
Inhalation	If the victim is not breathing, perform mouth-to-mouth resuscitation. Loosen tight clothing such as a collar, tie, belt or waistband. If breathing is difficult, oxygen can be administered. Seek medical attention if respiration problems do not improve.
Ingestion	INDUCE VOMITING by sticking finger in throat. Lower the head so that the vomit will not reenter the mouth and throat. Loosen tight clothing such as a collar, tie, belt or waistband. If the victim is not breathing, perform mouth-to-mouth resuscitation. Examine the lips and mouth to ascertain whether the tissues are damaged, a possible indication that the toxic material was ingested; the absence of such signs, however, is not conclusive.

Section V. Fire and Explosion Data					
Flammability	May be combustible at high temperature.	Auto-Ignition	Not available.		
Flash Points	112°C (233.6°F).	Flammable Limits	Not available.		
Combustion Products	These products are toxic carbon oxides (CO, CO ₂).				
Fire Hazards	Not available.				
Explosion Hazards	Risks of explosion of the product in presence of mechanical impact: Not available. Risks of explosion of the product in presence of static discharge: Not available.				
Fire Fighting Media and Instructions	SMALL FIRE: Use DRY chemical powder. LARGE FIRE: Use water spray, fog or foam. DO NOT use water jet. Consult with local fire authorities before attempting large scale fire-fighting operations.				

Section VI. Accidental Release Measures

Spill Cleanup Instructions

Absorb with an inert material and put the spilled material in an appropriate waste disposal. Finish cleaning the spill by rinsing any contaminated surfaces with copious amounts of water. Consult federal, state, and/or local authorities for assistance on disposal.

Section VII. Handling and Storage

Handling and Storage Information

Keep away from heat. Mechanical exhaust required. When not in use, tightly seal the container and store in a dry, cool place. Avoid excessive heat and light. Do not breathe gas/fumes/ vapor/spray

Section VIII. Exposure Controls/Personal Protection

Engineering Controls

Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapors below their respective threshold limit value. Ensure that eyewash station and safety shower is proximal to the work-station location.

Personal Protection

Splash goggles. Lab coat. Vapor respirator. Boots. Gloves. Suggested protective clothing might not be sufficient; consult a specialist BEFORE handling this product. Be sure to use a MSHA/NIOSH approved respirator or equivalent.



Exposure Limits

Not available

Section IX. Physical and Chemical Properties					
Physical state @ 20°C	Liquid. (Clear, Light Yellow.)	Solubility	Not available.		
Specific Gravity	1 (water=1)	_			
Molecular Weight	156.22	Partition Coefficient	Not available.		
Boiling Point	263 to 264 °C (505.4 to 507.2 °F)	Vapor Pressure	Not available.		
Melting Point	-17 to -16 °C (1.4 to 3.2 °F)	Vapor Density	Not available.		
Refractive Index	1.608	Volatility	Not available.		
Critical Temperature	Not available.	Odor	Not available.		
Viscosity	Not available.	Taste	Not available.		

Section X. Stability and Reactivity Data

Stability

This material is stable if stored under proper conditions. (See Section VII for instructions)

Conditions of Instability

Avoid excessive heat and light.

Incompatibilities Reactive with strong oxidizing agents.

Section XI. Toxicological Information

RTECS Number

Not available

Routes of Exposure

Eye Contact. Ingestion. Inhalation.

Toxicity Data

Not available.

Chronic Toxic Effects

CARCINOGENIC EFFECTS: Not available. MUTAGENIC EFFECTS : Not available TERATOGENIC EFFECTS : Not available. **DEVELOPMENTAL TOXICITY**: Not available.

Repeated or prolonged exposure to this compound is not known to aggravate existing medical conditions.

Acute Toxic Effects

No specific information is available in our data base regarding the toxic effects of this material for humans. However, exposure to any chemical should be kept to a minimum. Skin and eye contact may result in irritation. May be harmful if inhaled or ingested. Always follow safe industrial hygiene practices and wear proper protective equipment when handling

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Section XII. Ecological Information

Ecotoxicity

Not available.

Environmental Fate

1,6-Dimethylnaphthalene's formation during petroleum refining and coal tar distillation, as a result of gasoline, diesel, and heating fuel combustion, and from municipal waste incinerator activities may result in its direct release to the environment. The compound is a natural component of crude oil. If released to air, a vapor pressure of 1.46X10-2 mm Hg at 25 deg C indicates 1,6-dimethylnaphthalene will exist primarily as a vapor in the ambient atmosphere. Vapor-phase 1,6-dimethylnaphthalene will be degraded in the atmosphere by reaction with photochemically-produced hydroxyl radicals; the half-life for this reaction in air is estimated to be 5 hrs. Based upon aqueous photolysis data for 1- and 2-methylnaphthalene, 1,6-dimethylnaphthalene may undergo direct photolysis in the atmosphere. If released to soil, 1,6-dimethylnaphthalene is expected to have slight to no mobility based upon an estimated Koc of 4,900. Volatilization from moist soil surfaces is expected to be an important fate process based upon an estimated Henry's Law constant of 4.2X10-4 atm-cu m/mole. A 54.5% loss after 7 days in a marine water die-away study with sediment inoculum from Dunstaffnage Bay, Scotland, indicates that biodegradation may be an important fate process in the environment. If released into water, 1,6-dimethylnaphthalene is expected to adsorb to suspended solids and sediment based upon the estimated Koc. Volatilization from water surfaces is expected to be an important fate process based upon this compound's estimated Henry's Law constant. Estimated volatilization half-lives for a model river and model lake are 6 hrs and 6 days, respectively. However, volatilization from water surfaces is expected to be attenuated by adsorption to suspended solids and sediment in the water column. An estimated BCF of 520 suggests the potential for bioconcentration in aquatic organisms is high. Hydrolysis is not expected to be an important environmental fate process since this compound lacks functional groups that hydrolyze under environmental conditions. Occupational exposure to 1,6-dimethylnaphthalene may occur through inhalation and dermal contact with this compound at workplaces where 1,6-dimethylnaphthalene is produced. Monitoring data indicate that the general population may be exposed to 1,6-dimethylnaphthalene via ingestion of smoked foods and contaminated drinking water, and via inhalation of vehicle exhaust.

Section XIII. Disposal Considerations

Waste Disposal

Recycle to process, if possible. Consult your local regional authorities. You may be able to dissolve or mix material with a combustible solvent and burn in a chemical incinerator equipped with an afterburner and scrubber system. Observe all federal, state and local regulations when disposing of the substance.

Section XIV. Transport Information

DOT Classification

Not a DOT controlled material (United States).

PIN Number

Not applicable.

Proper Shipping Name

Not applicable.

Packing Group (PG)

Not applicable.

DOT Pictograms



Section XV. Other Regulatory Information and Pictograms

TSCA Chemical Inventory

(EPA)

This product is **NOT** on the EPA Toxic Substances Control Act (TSCA) inventory. The following notices are required by 40 CFR 720.36 (C) for those products not on the inventory list:

(i) These products are supplied solely for use in research and development by or under the supervision of a technically qualified individual as defined in 40 CFR 720.0 et sec.

(ii) The health risks of these products have not been fully determined. Any information that is or becomes available will be supplied on an MSDS sheet.

WHMIS Classification (Canada)

Not available.

EINECS Number (EEC)

209-385-1

EEC Risk Statements

Not available.

Japanese Regulatory Data

ENCS No. 4-80

Section XVI. Other Information

Version 1.0 Validated on 11/1/2005. Printed 11/1/2005.

Notice to Reader

TCI laboratory chemicals are for research purposes only and are NOT intended for use as drugs, food additives, households, or pesticides. The information herein is believed to be correct, but does not claim to be all inclusive and should be used only as a guide. Neither the above named supplier nor any of its subsidiaries assumes any liability whatsoever for the accuracy or completeness of the information contained herein. Final determination of suitability of any material is the sole responsibility of the user. All chemical reagents must be handled with the recognition that their chemical, physiological, toxicological, and hazardous properties have not been fully investigated or determined. All chemical reagents should be handled only by individuals who are familiar with their potential hazards and who have been fully trained in proper safety, laboratory, and chemical handling procedures. Although certain hazards are described herein, we can not guarantee that these are the only hazards which exist. Our MSDS sheets are based only on data available at the time of shipping and are subject to change without notice as new information is obtained. Avoid long storage periods since the product is subject to degradation with age and may become more dangerous or hazardous. It is the responsibility of the user to request updated MSDS sheets for products that are stored for extended periods. Disposal of unused product must be undertaken by qualified personnel who are knowledgeable in all applicable regulations and follow all pertinent safety precautions including the use of appropriate protective equipment (e.g. protective goggles, protective clothing, breathing equipment, facial mask, fume hood). For proper handling and disposal, always comply with federal, state, and local regulations.