



Material Safety Data Sheet

HAZARD WARNINGS









Flammable material; avoid heat and sources of ignition. Toxic compound, do not ingest or inhale. Avoid all contact with this material.

RISK PHRASES

Irritating to skin, eyes, and the respiratory system. Aspiration hazard; may be harmful if swallowed and enters airwavs

POSSIBLE MUTAGEN. MINIMIZE EXPOSURE.

Reproductive effects; may damage fertility or the unborn child. Environmental hazard. This material is toxic to aquatic organisms and may cause long term adverse effects to the aquatic environment.

Heat sensitive material. May develop pressure.

PROTECTIVE CLOTHING







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Section I. C	hemical Product and Company Id	lentification	
Chemical Name	Carbon Disulfide		
Catalog Number	C1955	Supplier	TCI America 9211 N. Harborgate St.
Synonym	Not available.		Portland OR 1-800-423-8616
Chemical Formula	$\overline{ ext{CS}_2}$	· · · · · · · · · · · · · · · · · · ·	
CAS Number	75-15-0	In case of Emergency Call	Chemtrec® (800) 424-9300 (U.S.) (703) 527-3887 (International)
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Section II. Composition as	nd Informa	tion on Ing	gredients	
Chemical Name	CAS Number	Percent (%)	TLV/PEL	Toxicology Data
Carbon Disulfide	75-15-0	,	This compound is classified as a possible mutagen. There is no acceptable exposure limit for a mutagen.	

Section III. Hazards Identification

Acute Health Effects Toxic if ingested or inhaled. Avoid prolonged contact with this material. Overexposure may result in serious illness or death.

Irritating to eyes and skin on contact. Inhalation causes irritation of the lungs and respiratory system. Inflammation of the eye is characterized by redness, watering, and itching. Skin inflammation is characterized by itching, scaling, reddening, or, occasionally, blistering.

Follow safe industrial hygiene practices and always wear proper protective equipment when handling this compound.

Chronic Health Effects CARCINOGENIC EFFECTS: Not available.

MUTAGENIC EFFECTS: Not available. TERATOGENIC EFFECTS: Not available.

DEVELOPMENTAL TOXICITY: Reproductive effects.

Man TCLo Inhalation 40 mg/m³, male 91 weeks prior to mating

Toxic Effects:

Paternal Effects - Spermatogenesis

Rat TCLo Inhalation 100/mg3 for 8 hours, female 1-21 days of pregnancy

Toxic Effects:

Effects on Embryo or Fetus - Fetal death

Specific Developmental Abnormalities - Craniofacial Specific Developmental Abnormalities - Homeostasis

Mouse TCLo Inhalation 2000 mg/m³ for 2 hours, female 1-21 days of pregnancy

Toxic Effects:

Effects on Fertility - Pre-implantation mortality

Effects on Fertility - Litter size

Repeated exposure to an highly toxic material may produce general deterioration of health by an accumulation in one or

many human organs.

Section IV.	First Aid Measures
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Check for and remove any contact lenses. In case of contact, immediately flush eyes with plenty of water for at least 15 Eve Contact minutes. Get medical attention.

In case of contact, immediately flush skin with plenty of water for at least 15 minutes while removing contaminated clothing Skin Contact and shoes. Wash clothing before reuse. Thoroughly clean shoes before reuse. Get medical attention immediately.

If the victim is not breathing, perform mouth-to-mouth resuscitation. Loosen tight clothing such as a collar, tie, belt or

waistband. If breathing is difficult, oxygen can be administered. Seek medical attention if respiration problems do not

INDUCE VOMITING by sticking finger in throat. Lower the head so that the vomit will not reenter the mouth and throat. Ingestion Loosen tight clothing such as a collar, tie, belt or waistband. If the victim is not breathing, perform mouth-to-mouth resuscitation. Examine the lips and mouth to ascertain whether the tissues are damaged, a possible indication that the toxic material was ingested; the absence of such signs, however, is not conclusive.

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Inhalation

Emergency phone number (800) 424-9300

C1955	C	arbon Disulfide	Page 2
Section V.	Fire and Explosion Data		
Flammability	Flammable.	Auto-Ignition	90 to 100 °C (194 to 212 °F)
Flash Points	-30 °C (-22 °F)	Flammable Limits	LOWER: 1% UPPER: 50%
Combustion Products	These products include toxic carbon oxides (C	CO,CO ₂), sulfur oxides (SO _x).	
Fire Hazards	Not available.		
Explosion Hazards	Risks of explosion of the product in presence of Risks of explosion of the product in presence of		
Fire Fighting Media and Instructions	Flammable liquid. SMALL FIRE: Use DRY chemical powder. LARGE FIRE: Use alcohol foam, water spray Consult with local fire authorities before attempt		perations.
Section VI.	Accidental Release Measures		
Spill Cleanup Instructions	Environmentally hazardous material. Heat sen Keep away from heat. Mechanical exhaust non-combustible material. DO NOT get wate	sitive material. required. Stop leak if withor rinside container. DO NOT to	essibly mutagenic material. Reproductive effector. but risk. Absorb with DRY earth, sand or other buch spilled material. Use water spray to reduce if needed. Consult federal, state, and/or local
Section VII.	Handling and Storage		
Handling and Storage Information	ENVIRONMENTAL HAZARD. HEAT SENSI from heat. Mechanical exhaust required. A	TIVE MATERIAL. MAY DEVE void excessive heat and ligh: g. If ingested, seek medical a	E MUTAGEN. REPRODUCTIVE EFFECTOR. LOP PRESSURE. Keep locked up. Keep away t. DO NOT ingest. Do not breathe gas/fumes/dvice immediately and show the container or the metals.
Section VIII.	Exposure Controls/Personal F	Protection	
Engineering Controls	Provide exhaust ventilation or other engineering threshold limit value. Ensure that eyewash states		e concentrations of vapors below their respective imal to the work-station location.
Personal Protection			OSH approved respirator must be used to avoid icient; consult a specialist BEFORE handling this
Personal Protection	inhalation of the product. Suggested protect		
Personal Protection Exposure Limits	inhalation of the product. Suggested protect	tive clothing might not be suff	icient; consult a specialist BEFORE handling this
Exposure Limits	inhalation of the product. Suggested protect product.	etive clothing might not be suff	icient; consult a specialist BEFORE handling this
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Exposure Limits Section IX. Physical state @ 20°C	inhalation of the product. Suggested protect product. This compound is classified as a possible mut Physical and Chemical Proper Liquid. (Clear, colorless ~ very pale yellow.)	agen. There is no acceptable	exposure limit for a mutagen. Miscible with ethanol, ether, benzene, chloroform, carbon tetrachloride, anhydrous methanol, oils. Very slightly soluble in water
Exposure Limits Section IX. Physical state @ 20°C Specific Gravity	inhalation of the product. product. This compound is classified as a possible mut Physical and Chemical Proper Liquid. (Clear, colorless ~ very pale yellow.) 1.26 (water=1)	agen. There is no acceptable ties Solubility	exposure limit for a mutagen. Miscible with ethanol, ether, benzene, chloroform, carbon tetrachloride, anhydrous methanol, oils. Very slightly soluble in water (0.2g/100mL 20 °C).
Exposure Limits Section IX. Physical state @ 20°C Specific Gravity Molecular Weight	inhalation of the product. product. This compound is classified as a possible mut Physical and Chemical Proper Liquid. (Clear, colorless ~ very pale yellow.) 1.26 (water=1) 76.14	agen. There is no acceptable rties Solubility Partition Coefficient	exposure limit for a mutagen. Miscible with ethanol, ether, benzene, chloroform, carbon tetrachloride, anhydrous methanol, oils. Very slightly soluble in water (0.2g/100mL 20°C). LOG P _{ow} : 2.11
Exposure Limits Section IX. Physical state @ 20°C Specific Gravity Molecular Weight Boiling Point	inhalation of the product. product. This compound is classified as a possible mut Physical and Chemical Proper Liquid. (Clear, colorless ~ very pale yellow.) 1.26 (water=1) 76.14 46℃ (114.8℃)	agen. There is no acceptable ties Solubility Partition Coefficient Vapor Pressure	exposure limit for a mutagen. Miscible with ethanol, ether, benzene, chloroform, carbon tetrachloride, anhydrous methanol, oils. Very slightly soluble in water (0.2g/100mL 20°C). LOG P _{ow} : 2.11 48 kPa (@ 25°C)
Exposure Limits Section IX. Physical state @ 20°C Specific Gravity Molecular Weight Boiling Point Melting Point	inhalation of the product. product. This compound is classified as a possible mut Physical and Chemical Proper Liquid. (Clear, colorless ~ very pale yellow.) 1.26 (water=1) 76.14 46℃ (114.8℉) -111℃ (-167.8℉)	agen. There is no acceptable ties Solubility Partition Coefficient Vapor Pressure Vapor Density	exposure limit for a mutagen. Miscible with ethanol, ether, benzene, chloroform, carbon tetrachloride, anhydrous methanol, oils. Very slightly soluble in water (0.2g/100mL 20°C). LOG P _{ow} : 2.11 48 kPa (@ 25°C) 2.63 (Air = 1)
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Section XI. Toxicological Information

FF6650000 RTECS Number

Routes of Exposure

Eye Contact. Ingestion. Inhalation.

Toxicity Data

Rat LD₅₀ (oral) 1200 mg/kg Mouse LD₅₀ (oral) 2780 mg/kg Rat LD₅₀ (inhalation) 25 gm/m³/2H

Chronic Toxic Effects

CARCINOGENIC EFFECTS: Not available. MUTAGENIC EFFECTS: Not available. TERATOGENIC EFFECTS: Not available.

DEVELOPMENTAL TOXICITY: Reproductive effects. Man TCLo Inhalation 40 mg/m³, male 91 weeks prior to mating

Paternal Effects - Spermatogenesis

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Toxic Effects:

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Specific Developmental Abnormalities - Craniofacial Specific Developmental Abnormalities - Homeostasis

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Toxic Effects:

Effects on Fertility - Pre-implantation mortality Effects on Fertility - Litter size

Repeated exposure to an highly toxic material may produce general deterioration of health by an accumulation in one or many

human organs.

Acute Toxic Effects

Toxic if ingested or inhaled. Avoid prolonged contact with this material. Overexposure may result in serious illness or death. Irritating to eyes and skin on contact. Inhalation causes irritation of the lungs and respiratory system. Inflammation of the eye is characterized by redness, watering, and itching. Skin inflammation is characterized by itching, scaling, reddening, or, occasionally, blistering.
Follow safe industrial hygiene practices and always wear proper protective equipment when handling this compound.

Section XII. **Ecological Information**

Ecotoxicity

Not available.

Environmental Fate

Carbon disulfide's production and use as a solvent and a chemical intermediate may result in its release to the environment through various waste streams. Its former production and use as an insecticide used for fumigation of nursery stock and soil treatment against insects and nematodes, resulted in its direct release to the environment. Carbon disulfide is a natural product of anaerobic biodegradation and is released to the atmosphere from oceans and landmasses as well as geothermal sources. The ocean appears to be a major source of carbon disulfide. Coastal and marshland areas of high biological activity are also a major source. If released to air, a vapor pressure of 359 mm Hg at 25 deg C indicates carbon disulfide will exist solely as a vapor in the ambient atmosphere. Vapor-phase carbon disulfide will be degraded in the atmosphere by reaction with photochemically-produced hydroxyl radicals; the half-life for this reaction in air is estimated to be 5.5 days. Carbon disulfide has a weak UV adsorption band at 317 nm, suggesting a potential for direct photolysis. However, photolysis is not considered to be a relevant loss mechanism for the chemical. If released to soil, carbon disulfide is expected to have moderate mobility based upon an estimated Koc of 270. Volatilization from moist soil surfaces is expected to occur based upon a Henry's Law constant of 1.44X10-2 atm-cu m/mole at 24 deg C. Carbon disulfide may potentially volatilize from dry soil surfaces given its vapor pressure. If released into water, carbon disulfide is not expected to adsorb to suspended solids and sediment in the water column based upon the estimated Koc. Volatilization from water surfaces is expected to be an important fate process based upon carbon disulfide's Henry's Law constant. Estimated volatilization half-lives for a model river and model lake are 2.6 hours and 3.5 days, respectively. BCFs of <6.1 and <60 in carp suggest bioconcentration in aquatic organisms is low to moderate. Carbon disulfide hydrolyzes slowly to carbon dioxide and hydrogen disulfide in alkaline solutions. The half-life for hydrolysis at pH 9 is approximately 1.1 years. Occupational exposure to carbon disulfide may occur through inhalation and dermal contact with this compound at workplaces where carbon disulfide is produced or used. As carbon disulfide occurs ubiquitously in the environment, the general population is exposed to this compound. Primary routes of exposure to carbon disulfide are through inhalation of ambient air or ingestion of fruits, vegetables, and other food products containing this compound.

Section XIII. Disposal Considerations

Waste Disposal

Recycle to process, if possible. Consult your local regional authorities. You may be able to dissolve or mix material with a combustible solvent and burn in a chemical incinerator equipped with an afterburner and scrubber system. Observe all federal, state and local regulations when disposing of the substance.

Section XIV. Transport Information

DOT Classification

DOT CLASS 3: Flammable liquid DOT CLASS 6.1: Toxic material

PIN Number

UN1131

Proper Shipping Name

Carbon disulfide

Packing Group (PG)

RQ = 100 (45.4)

DOT Pictograms





Emergency phone number (800) 424-9300

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Section XV. Other Regulatory Information and Pictograms

TSCA Chemical Inventory (EPA)

This compound is ON the EPA Toxic Substances Control Act (TSCA) inventory list.

WHMIS Classification

CLASS B-2: Flammable liquid with a flash point lower than 37.8 °C (100 °F). CLASS D-1B: Material causing immediate and serious toxic effects (TOXIC). (Canada)

On DSI

EINECS Number (EEC)

200-843-6

EEC Risk Statements R11- Highly flammable.

R18- In use, may form flammable/explosive vapor-air mixture. R23/24/25- Toxic by inhalation, in contact with skin and if swallowed.

R36/37/38- Irritating to eyes, respiratory system and skin.

R46- May cause heritable genetic damage.

R47- May cause birth defects.

R51- Toxic to aquatic organisms.

R53- May cause long-term adverse effects in the aquatic environment.

R62- Possible risk of impaired fertility. R63- Possible risk of harm to unborn child.

Japanese Regulatory Data

ENCS No. 1-172

Section XVI. Other Information

Version 1.0 Validated on 3/21/2011. Printed 3/21/2011.

Notice to Reader

TCI laboratory chemicals are for research purposes only and are NOT intended for use as drugs, food additives, households, or pesticides. The information herein is believed to be correct, but does not claim to be all inclusive and should be used only as a guide. Neither the above named supplier nor any of its subsidiaries assumes any liability whatsoever for the accuracy or completeness of the information contained herein. Final determination of suitability of any material is the sole responsibility of the user. All chemical reagents must be handled with the recognition that their chemical, physiological, toxicological, and hazardous properties have not been fully investigated or determined. All chemical reagents should be handled only by individuals who are familiar with their potential hazards and who have been fully trained in proper safety, laboratory, and chemical handling procedures. Although certain hazards are described herein, we can not guarantee that these are the only hazards which exist. Our MSDS sheets are based only on data available at the time of shipping and are subject to change without notice as new information is obtained. Avoid long storage periods since the product is subject to degradation with age and may become more dangerous or hazardous. It is the responsibility of the user to request updated MSDS sheets for products that are stored for extended periods. Disposal of unused product must be undertaken by qualified personnel who are knowledgeable in all applicable regulations and follow all pertinent safety precautions including the use of appropriate protective equipment (e.g. protective goggles, protective clothing, breathing equipment, facial mask, fume hood). For proper handling and disposal, always comply with federal, state, and local regulations.

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